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in a group 2 element c the electron for the first ionization energy is removed from a spin pair in an s sublevel 1 for d a group 13 element the electron for the first ionization energy is removed from a 3 p sublevel further away from

the nucleus 1 the 3 p sublevel also experiences more shielding

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3 a 1.20×10^{24} atoms of fe b 2.63×10^{24} atoms of si c 7.23×10^{21} atoms of li d 5.27×10^{23} s²⁻ ions 4 a 5.72×10^{22} molecules of water h₂o b 1.93×10^{21} formula units of sodium chloride nacl c 1.24×10^{24} formula units of calcium bromide cabr₂ d 8.43×10^{22} molecules of glucose c₆h₁₂o

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solid potassium chlorate KClO_3 decomposes to form solid potassium chloride and diatomic oxygen gas
solid aluminum metal reacts with solid diatomic iodine to form solid Al_2I_6

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