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Chemistry 'O' Level Rex M. Heyworth,2007

Descriptive Inorganic Chemistry James E. House,Kathleen A. House,2010-09-22 Descriptive Inorganic Chemistry, Second Edition, covers the synthesis, reactions, and properties of elements and inorganic compounds for courses in descriptive inorganic chemistry. This updated version includes expanded coverage of chemical bonding and enhanced treatment of Buckminster Fullerenes, and incorporates new industrial applications matched to key topics in the text. It is suitable for the one-semester (ACS-recommended) course or as a supplement in general chemistry courses. Ideal for majors and non-majors, the book incorporates rich graphs and diagrams to enhance the content and maximize learning. - Includes expanded coverage of chemical bonding and enhanced treatment of Buckminster Fullerenes - Incorporates new industrial applications matched to key topics in the text

Educational Times ,1899

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General Chemistry Ralph H. Petrucci,2007 General Chemistry: Principles and Modern Applicationsis recognized for its superior problems, lucid writing, precision of argument, and precise and detailed treatment of the subject. Popular and innovative features include Feature Problems, follow-up A and B Practice Exercises to accompany every in-chapter Example,

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Sif: Chemistry 5na Tb J. G. R. Briggs, 2009

Inorganic Chemistry James E. House, 2019-11-01 Inorganic Chemistry, Third Edition, emphasizes fundamental principles, including molecular structure, acid-base chemistry, coordination chemistry, ligand field theory and solid state chemistry. The book is organized into five major themes: structure, condensed phases, solution chemistry, main group and coordination compounds, each of which is explored with a balance of topics in theoretical and descriptive chemistry. Topics covered include the hard-soft interaction principle to explain hydrogen bond strengths, the strengths of acids and bases, and the stability of coordination compounds, etc. Each chapter opens with narrative introductions and includes figures, tables and end-of-chapter problem sets. This new edition features updates throughout, with an emphasis on bioinorganic chemistry and a new chapter on nanostructures and graphene. In addition, more in-text worked-out examples encourage active learning and prepare students for exams. This text is ideal for advanced undergraduate and graduate-level students enrolled in the Inorganic Chemistry course. - Includes physical chemistry to show the relevant principles from bonding theory and thermodynamics - Emphasizes the chemical characteristics of main group elements and coordination chemistry - Presents chapters that open with narrative introductions, figures, tables and end-of-chapter problem sets

The Educational Times, and Journal of the College of Preceptors, 1899

Organic Chemistry L. G. Wade, 2006

Subject Guide to Books in Print, 1984

Readings in Earth Science and Physical Science Regina Avraham, 1988-03

Virtual Chemlab Brian F. Woodfield, Matthew C. Asplund, Steven Haderlie, 2006 Contains a full virtual lab environment as well as the pre-arranged labs that are referenced in the workbook and at the end of the chapter in the textbook. Virtual ChemLab can be run directly from the CD or installed on the student's computer.

Commerce Business Daily, 2001-12-03

English Mechanics and the World of Science, 1871

The Software Encyclopedia, 1988

Druggists' Circular, 1922

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